

Table 23.10 Values of  $pK_a$  for selected pharmaceutical bases

Base	Cation	$pK_a$	Example
Potassium Hydroxide	Potassium	~14	Benzylpenicillin
Sodium hydroxide	Sodium	~14	Diclofenac
Zinc hydroxide	Zinc	~14	Bacitracine
Calcium hydroxide	Calcium	12.6, 11.57	Fenoprofen
Magnesium hydroxide	Magnesium	11.4	Menbutone
Choline	Choline	>11	Theophylline
Lysine	Lysine	10.79, 9.18, 2.16	Ibuprofen
Benzathine	Benzathine	9.99, 9.39	Ampicillin
Piperazine	Piperazine	9.82, 5.58	Naproxen
Meglumine	Meglumine	9.5	Flunixin
Ammonia	Ammonium	9.27	Glycyrrhizinic acid
Tromethamine	Trometamol	8.02	Lodoxamide
Aluminium hydroxide	Aluminium	>7	

(Stahl and Wermuth, 2011).

Table 23.11 Frequency of pharmaceutical anions and cations of drugs in USP 29-NF24

Anion	Frequency (%)	Cation	Frequency (%)
Hydrochloride	39.96	Sodium	62.79
Sulphate	10.58	Potassium	11.05
Acetate	6.70	Calcium	8.72
Phosphate	4.97	Aluminium	4.65
Chloride	4.54	Benzathine	2.33
Maleate	3.67	Meglumine	2.33
Citrate	3.02	Zinc	2.33
Mesilate	2.59	Magnesium	1.74
Succinate	2.38	Tromethamine	1.74
Nitrate	2.38	Lysine	1.16

metabolites in biochemical pathways. These include hydrochloride and sodium salts and, as such, they are considered to be unrestricted in their use.

*Second class* salt formers are those that are not naturally occurring but which have found common application and have not shown significant

toxicological or tolerability issues (such as the sulphonic acids, e.g. mesilates).

*Third class* salt formers are those that are used in special circumstances to solve a particular problem. They are not naturally occurring, nor are they in common use.