



FIG. 3.7 Infrared spectra of P3C2 glass and P3C3 ceramic glass versus soaking time in SBF.

- (5) The material obtained was then heat-treated at low temperature (250°C) in order to remove the PMMA and then sintered at 900°C for 3 h with a heating rate of 3°C/min.
- (6) The macroporous glass-ceramic presents walls and cavities with a relative density of 95%. The size of the interconnections is about 130 μm, and the mean diameter of the macropores is 490 μm, the pore volume being about 67%.

An X-ray diffraction analysis, as a function of the temperature, revealed that the glass crystallized from 500°C, predominantly in  $\text{Na}_2\text{CaSi}_3\text{O}_8$ . However, NMR analyses of  $^{31}\text{P}$  and  $^{29}\text{Si}$  on this glass, treated at 900°C, showed that a vitreous part remained in this material, and that this glass could be assimilated to a glass-ceramic.

In order to verify the application of this material as a bone substitute, tests of *in vitro* bioactivity were carried out on the initial glass P3C2 and on the glass-ceramic obtained from the P3C2 glass.

Infrared analyses (Fig. 3.7) on the surfaces of samples after soaking in SBF show that the glass-ceramic develops its HCA layer more quickly than glass. Indeed, the apatite was formed after 10 h 15 min soaking for the glass and only 5 h 15 min for the glass-ceramic.

### 3.3 MIXTURE DESIGNS APPLIED TO THE $\text{SiO}_2\text{-CaO-Na}_2\text{O}$ SYSTEM DOPED WITH $\text{P}_2\text{O}_5$ AND $\text{CaF}_2$

#### 3.3.1 Selection of Doping

Taking into account the differing bioactivity results of the  $\text{SiO}_2$ ,  $\text{CaO}$ , and  $\text{Na}_2\text{O}$ -based glasses obtained by various authors, we decided to apply a mixture design