

Calculate the percentage of 2-pyrrolidinone in the sample taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

- $r_U$  = peak response of 2-pyrrolidinone from the *Sample solution*  
 $r_S$  = peak response of 2-pyrrolidinone from the *Standard solution*  
 $C_S$  = concentration of 2-pyrrolidinone in the *Standard solution* (mg/mL)  
 $C_U$  = concentration of Povidone in the *Sample solution* (mg/mL), calculated on the anhydrous basis

Acceptance criteria: NMT 3.0%

#### • PEROXIDES

**Sample solution:** 40 mg/mL of Povidone in water, calculated on the anhydrous basis

**Blank:** To 25 mL of the *Sample solution*, add 2 mL of 13% sulfuric acid.

#### Instrumental conditions

(See *Ultraviolet-Visible Spectroscopy* (857).)

**Mode:** UV-Vis

**Analytical wavelength:** 405 nm

**Cell:** 1 cm

#### Analysis

**Sample:** *Sample solution*

To 25 mL of the *Sample solution*, add 2 mL of titanium trichloride-sulfuric acid TS, and allow to stand for 30 min. Measure the absorbance of a portion of this solution against the *Blank*.

Acceptance criteria: NMT 0.35, corresponding to NMT 400 ppm, expressed as H<sub>2</sub>O<sub>2</sub>

#### • FORMIC ACID

**Mobile phase:** Diluted perchloric acid (1 in 700)

**Standard solution:** 10 µg/mL of formic acid in water

**Sample stock solution:** 20 mg/mL of Povidone in water

**Sample solution:** Transfer a suspension of strongly acidic ion-exchange resin (use the hydrogen form of ion-exchange resin) in water to a column of about 8 mm in inside diameter to give a packing depth of about 20 mm in length. Keep the strongly acidic ion-exchange resin layer constantly immersed in water. Pour 5 mL of water and adjust the flow rate so that water drops at a rate of about 1 mL/min. When the level of the water is near the top of the strongly acidic ion-exchange resin layer, introduce 100 mL of the *Sample stock solution* into the column. Disregard the first 2 mL of the eluate, then collect 1.5 mL of the solution, and use this as the *Sample solution*.

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 7.8-mm × 30-cm; 9-µm packing L17

**Column temperature:** 35°

**Flow rate:** 1.0 mL/min

[NOTE—The retention time of formic acid is about 8 min.]

**Injection volume:** 50 µL

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Column efficiency:** NLT 1000 theoretical plates for the formic acid peak

**Symmetry factor:** 0.5–1.5 for the formic acid peak

**Relative standard deviation:** NMT 2.0% of formic acid for six injections

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Record the chromatograms and measure the responses for the formic acid peak.

Calculate the percentage of formic acid in the sample taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

- $r_U$  = peak response of formic acid from the *Sample solution*  
 $r_S$  = peak response of formic acid from the *Standard solution*  
 $C_S$  = concentration of formic acid in the *Standard solution* (mg/mL)  
 $C_U$  = concentration of Povidone in the *Sample solution* (mg/mL), calculated on the anhydrous basis

Acceptance criteria: NMT 0.5%

#### SPECIFIC TESTS

##### • PH (791)

**Sample solution:** 50 mg/mL in water

**Acceptance criteria:** 3.0–5.0 for Povidone having a nominal K-value of 30 or less; 4.0–7.0 for Povidone having a nominal K-value greater than 30

##### • WATER DETERMINATION (921), Method I: NMT 5.0%

##### • K-VALUE

**Sample solution:** Weigh a quantity of undried Povidone, equivalent on the anhydrous basis, to the amount specified in Table 1.

Table 1

Nominal K-value	Quantity (g)
≤18	5.00
>18 to ≤95	1.00
>95	0.10

Dissolve it in 50 mL of water in a 100-mL volumetric flask, and dilute to volume. Allow to stand for 1 h.

#### Analysis

**Samples:** *Sample solution* and water

Determine the viscosity of the *Sample solution* and the water, using a capillary-tube viscometer (see *Viscosity—Capillary Methods* (911)), at 25 ± 0.2°.

Calculate the K-value of Povidone:

$$\text{Result} = \left[ \sqrt{300c \log z + (c + 1.5c \log z)^2} + 1.5c \log z - c \right] / (0.15c + 0.003c^2)$$

$c$  = weight, on the anhydrous basis, of the specimen tested in each 100.0 mL of solution (g)

$z$  = viscosity of the *Sample solution* relative to that of water

#### Acceptance criteria

**K-value of Povidone having a stated (nominal) K-value of NMT 15:** 85.0%–115.0% of the stated values

**K-value of Povidone having a stated K-value or a stated K-value range with an average of more than 15:** 90.0%–108.0% of the stated value or of the average of the stated range

#### ADDITIONAL REQUIREMENTS

- **♦PACKAGING AND STORAGE:** Preserve in tight containers.♦
- **♦LABELING:** Label it to state, as part of the official title, the K-value or K-value range of Povidone.

##### • ♦USP REFERENCE STANDARDS (11)

USP Povidone RS.♦