

Sorbic Acid

1 Nonproprietary Names

BP: Sorbic Acid
PhEur: Sorbic Acid
USP-NF: Sorbic Acid

2 Synonyms

Acidum sorbicum; (2-butenylidene) acetic acid; crotylidene acetic acid; E200; hexadienic acid; hexadienoic acid; 2,4-hexadienoic acid; 1,3-pentadiene-1-carboxylic acid; 2-propenylacrylic acid; (*E,E*)-sorbic acid; *Sorbistat*.

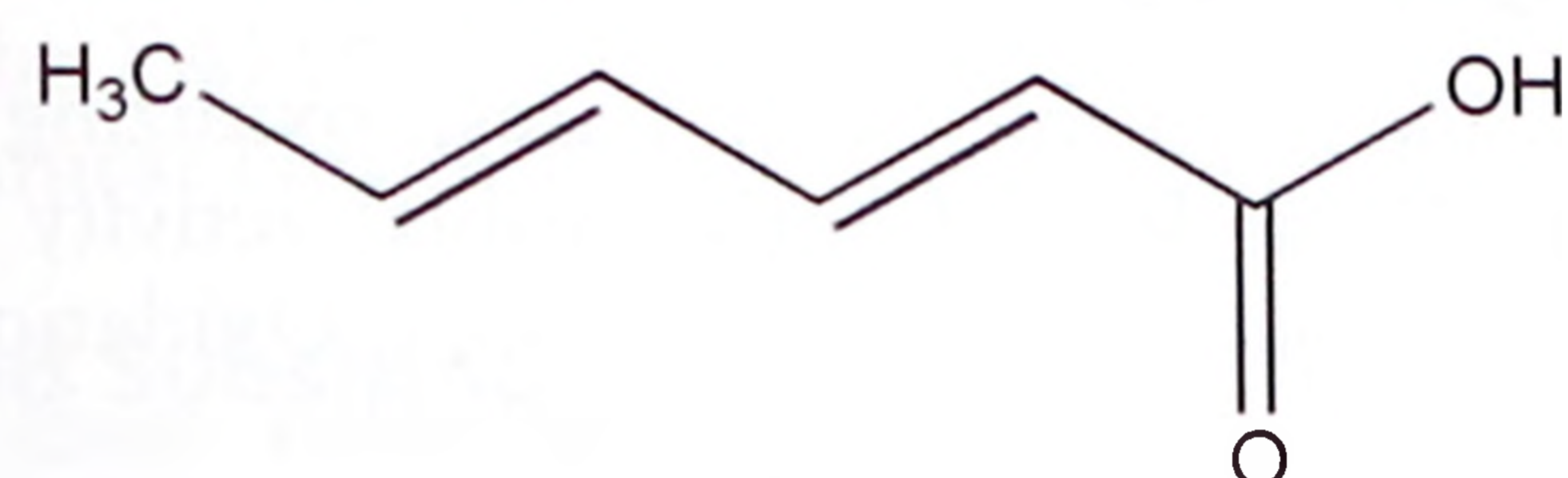
3 Chemical Name and CAS Registry Number

(*E,E*)-Hexa-2,4-dienoic acid [110-44-1] and [22500-92-1]

4 Empirical Formula and Molecular Weight

C₆H₈O₂ 112.13

5 Structural Formula



6 Functional Category

Antimicrobial preservative.

7 Applications in Pharmaceutical Formulation or Technology

Sorbic acid is an antimicrobial preservative^(1,2) with antibacterial and antifungal properties used in pharmaceuticals and enteral preparations. Generally, it is used at concentrations of 0.05–0.2% in oral and topical pharmaceutical formulations, especially those containing nonionic surfactants. Sorbic acid has been used in pluronic lecithin organogel formulations investigated for topical drug delivery⁽³⁾ and is also used with proteins, enzymes, gelatin, and vegetable gums.⁽⁴⁾ It has been shown to be an effective preservative for promethazine hydrochloride solutions in a concentration of 1 g/L.⁽⁵⁾

Sorbic acid is primarily used as an antifungal agent, although it also possesses antibacterial properties. The optimum antibacterial activity is obtained at pH 4.5; and practically no activity is observed above pH 6.^(6,7) Sorbic acid has limited stability and activity against bacteria and is thus frequently used in combination with other antimicrobial preservatives or glycols, when synergistic effects appear to occur.⁽⁸⁾

8 Description

Sorbic acid occurs as a tasteless, white to yellow-white crystalline powder with a faint characteristic odor.

9 Pharmacopeial Specifications

See Table I.

10 Typical Properties

Antimicrobial activity Reported minimum inhibitory concentrations (MICs) at pH 6 are shown in Table II.⁽⁹⁾

SEM 1: Excipient: sorbic acid; manufacturer: Pfizer Ltd.; magnification: 60×.



Table I: Pharmacopeial specifications for sorbic acid.

Test	PhEur 9.2	USP 40-NF 35 S1
Identification	+	+
Characters	+	—
Appearance of solution	+	—
Melting range	132–136°C	132–135°C
Water	≤1.0%	≤0.5%
Residue on ignition	—	≤0.2%
Sulfated ash	≤0.2%	—
Heavy metals	—	≤10 ppm
Aldehyde (as C ₂ H ₄ O)	≤0.15%	—
Assay (anhydrous basis)	99.0–101.0%	99.0–101.0%

Table II: Minimum inhibitory concentrations (MICs) of sorbic acid at pH 6.

Microorganism	MIC (μg/mL)
<i>Aspergillus niger</i>	200–500
<i>Candida albicans</i>	25–50
<i>Clostridium sporogenes</i>	100–500
<i>Escherichia coli</i>	50–100
<i>Klebsiella pneumoniae</i>	50–100
<i>Penicillium notatum</i>	200–300
<i>Pseudomonas aeruginosa</i>	100–300
<i>Pseudomonas cepacia</i>	50–100
<i>Pseudomonas fluorescens</i>	100–300
<i>Saccharomyces cerevisiae</i>	200–500
<i>Staphylococcus aureus</i>	50–100

Boiling point 228°C with decomposition.

Density 1.20 g/cm³

Dissociation constant pK_a = 4.76

Flash point 127°C

Melting point 134.5°C