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21 Author

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22 Date of Revision

4 May 2017.

Chlorobutanol

1 Nonproprietary Names

BP: Chlorobutanol

JP: Chlorobutanol

PhEur: Chlorobutanol, Anhydrous

USP-NF: Chlorobutanol

2 Synonyms

Acetone chloroform; anhydrous chlorbutol; chlorbutanol; chlorobutanolum anhydricum; chlorbutol; chloretone; *Coliquifilm*; methaform; sedaform; trichloro-*tert*-butanol; β,β,β -trichloro-*tert*-butyl alcohol; trichloro-*t*-butyl alcohol.

3 Chemical Name and CAS Registry Number

1,1,1-Trichloro-2-methyl-2-propanol [57-15-8]

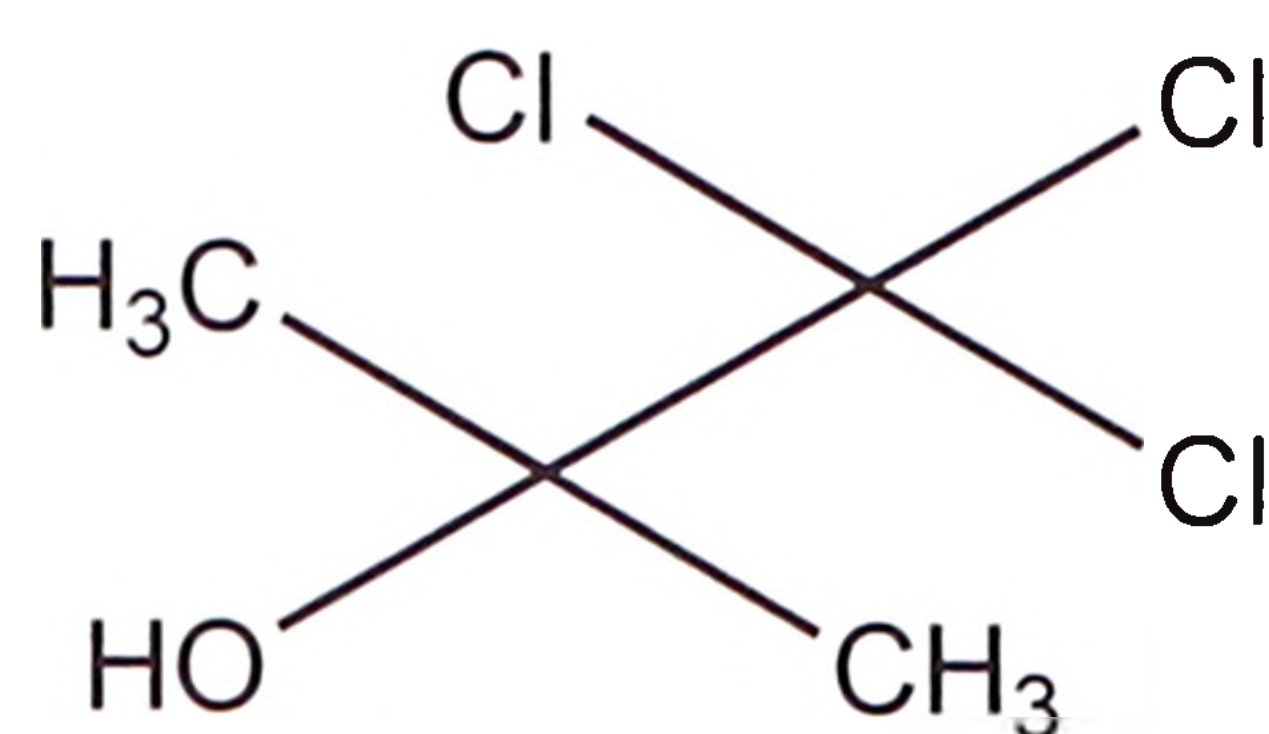
1,1,1-Trichloro-2-methyl-2-propanol hemihydrate [6001-64-5]

4 Empirical Formula and Molecular Weight

$C_4H_7Cl_3O$ 177.46 (for anhydrous)

$C_4H_7Cl_3O \cdot \frac{1}{2}H_2O$ 186.46 (for hemihydrate)

5 Structural Formula



6 Functional Category

Antimicrobial preservative; plasticizing agent.

7 Applications in Pharmaceutical Formulation or Technology

Chlorobutanol is primarily used in ophthalmic or parenteral dosage forms as an antimicrobial preservative at concentrations up to 0.5% w/v; *see* Section 10. It is commonly used as an antibacterial agent for epinephrine solutions, posterior pituitary extract solutions, and ophthalmic preparations intended for the treatment of miosis. It is especially useful as an antibacterial agent in nonaqueous formulations. Chlorobutanol is also used as a preservative in cosmetics (*see* Section 16); as a plasticizer for cellulose esters and ethers.

8 Description

Volatile, colorless or white crystals with a musty, camphoraceous odor.

9 Pharmacopeial Specifications

See Table I.

Table I: Pharmacopeial specifications for chlorobutanol.

Test	JP XVII	PhEur 9.2	USP 40-NF 35 S1
Identification	+	+	+
Characters	+	+	—
Appearance of solution	—	+	—
Melting point	$\geq 76^\circ C$	+	—
Anhydrous	—	$\approx 95^\circ C$	—
Hemihydrate	—	$\approx 78^\circ C$	—
Acidity	+	+	+
Water (anhydrous form)	$\leq 6.0\%$	$\leq 1.0\%$	$\leq 1.0\%$
Hemihydrate	—	4.5–5.5%	$\leq 6.0\%$
Chloride	$\leq 0.071\%$	+	$\leq 0.07\%$
Anhydrous	—	≤ 300 ppm	—
Hemihydrate	—	≤ 100 ppm	—
Residue on ignition	$\leq 0.10\%$	—	—
Sulfated ash	—	$\leq 0.1\%$	—
Assay (anhydrous basis)	$\geq 98.0\%$	98.0–101.0%	98.0–100.5%

Note: the JP XVII and USP 40-NF 35 S1 allow either the anhydrous form or the hemihydrate; the PhEur includes them as separate monographs.

10 Typical Properties

Antimicrobial activity Chlorobutanol has both antibacterial and antifungal properties. It is effective against Gram-positive and Gram-negative bacteria and some fungi, e.g. *Candida albicans*, *Pseudomonas aeruginosa*, and *Staphylococcus albus*. Antimicrobial activity is bacteriostatic, rather than bactericidal, and is considerably reduced above pH 5.5. In addition, activity may also be reduced by increasing heat and by incompatibilities between chlorobutanol and other excipients or packaging materials; *see* Sections 11 and 12. However, activity may be increased by combination with other antimicrobial preservatives; *see* Section 18. Typical minimum inhibitory concentrations (MICs) are: Gram-positive bacteria 650 $\mu g/mL$; Gram-negative bacteria 1000 $\mu g/mL$; yeasts 2500 $\mu g/mL$; fungi 5000 $\mu g/mL$.

Boiling point 167°C

Melting point

76–78°C for the hemihydrate;

95–97°C for the anhydrous form.

Partition coefficient Octanol: water $\log k_{ow} = 2.03$

Refractive index $n_D^{25} = 1.4339$

Solubility *see* Table II.